

BAZHANOVA, N.V.; MASLOVA, T.G.; POPOVA, I.A.; POPOVA, O.F.;  
SAPOZHNIKOV, D.I.; DYDEL'MAN, Z.M. Prinimali uchast'ye:  
CHERNOMORSKIY, S.M.; MENITSKAYA, I.M.; SAPOZHNIKOV, D.I.,  
otv. red.

[Plastid pigments of green plants and the methods of their  
study] Pigmenty plastid zelenykh rastenii i metodika ikh  
issledovaniia. Moskva, Izd-vo "Nauka," 1964. 119 p.

(MIRA 17:7)

1. Akademiya nauk SSSR. Botanicheskiy institut. 2. Labora-  
toriya fotosinteza Botanicheskogo instituta im. V.L.  
Komarova AN SSSR (for all except Sapozhnikov).

EYDEL'MAN, Z.M.; SAPOZHNIKOV, D.I.; BAZHANOVA, N.V.; MASLOVA, T.G.;  
POPOVA, O.F.; SHIRYAYEVA, G.A.

Relation between phosphorylation reactions and the transformation  
of xanthophylls in the course of photosynthesis. Trudy Bot. inst.  
Ser. 4 no.15:224-233 '62. (MIRA 15:7)  
(Xanthophyll) (Photosynthesis) (Phosphorylation)

SAPOZHNIKOV, D.I.; EYDEL'MAN, Z.M.; BAZHANOVA, N.V.; MASLOVA, T.G.;  
POPOVA, O.F.

Concerning the participation of carotenoids in the process of  
photosynthesis. Trudy Bot. inst. Ser. 4 no.15:43-52 '62.  
(MIRA 15:7)  
(Photosynthesis) (Carotenoids)

SAPOZHNIKOV, D.I.; MASLOVA, T.G.; BAZHANOVA, N.V.; POPOVA, O.F.;  
CHERNOMORSKIY, S.A.; SHIRYAYEVA, G.A.

State of pigments in leaves. Trudy Bot. inst. Ser. 4 no.15:  
53-67 '62. (MIRA 15:7)  
(Chlorophyll) (Carotenoids)

EYDEL'MAN, E.M.; POPOVA, O.F.

Methods for studying the photosynthetic phosphorylation. Trudy Otd.  
fiziol. i biofiz. rast. AN Tadzh. SSR 2:194-231 '62.

(MIRA 16:4)

(Photosynthesis)

(Phosphorylation)

EYDEL'MAN, Z.M.; SAPOZHNIKOV, D.I.; BAZHANOVA, N.V.; POPOVA, O.F.

Comparative study of the effect of photosynthetic poisons on  
photochemical conversion of some xanthophylls. Fiziol. rast 7  
no.2:129-132 '60. (MIRA 14:5)

1. Komarov Botanical Institute, U.S.S.R Academy of Sciences,  
Leningrad.

(Xanthophylls)  
(Photosynthesis)  
(Phosphorylation)

EYDEL'MAN, Z.M.; POPOVA, O.F.

Characteristics of the biosynthesis of plastid phytochromes in  
hybird and inbred corn forms[w.s.i.E.]. Trudy Bot. inst. Ser.4  
no.14:100-111 '60. (MIRA 14:3)  
(Corn breeding) (Chromatophores)

SAPOZHNIKOV, D.I.; ALKHAZOV, D.G.; EYDEL'MAN, Z.M.; BAZHANOVA, N.V.; LEMBERG, I.Kh.; MASLOVA, T.G.; GIRSHIN, A.B.; POPOVA, I.A.; SAAKOV, V.S.; POPOVA, O.F.; SHIRYAYEVA, G.A.

Incorporation of  $O^{18}$  from heavy oxygen water into violaxanthin due to the action of light on plants. Bot. zhur. 46 no. 5:673-676 My '61.  
(MIRA 14:7)

1. Botanicheskiy institut imeni V.L. Komarova AN SSSR, Leningrad.  
(Oxygen—Isotopes) (Violaxanthin)



SAPOZHNIKOV, D.I.; ALKHAZOV, D.G.; EYDEL'MAN, Z.M.; BAZHANOVA, N.V.;  
LEMBERG, I.Kh.; MASLOVA, T.G.; GINSHIN, A.B.; POPOVA, I.A.;  
SAAKOV, V.S.; POPOVA, O.F.; SHIRYAYEVA, G.A.

Participation of xanthophylls in oxygen transport in the  
process of photosynthesis. Dokl. AN SSSR 154 no.4:974-977  
P '64. (MIRA 17:3)

1. Botanicheskiy institut im. V.L. Komarova AN SSSR. Pred-  
stavleno akademikom A.L. Kursanovym.

SAPOZHNIKOV, D.I.; MASLOVA, T.G.; BAZHANOVA, N.V.; POPOVA, O.F.

Kinetics of the inclusion of  $O^{18}$  from heavy oxygen water into the violaxanthin molecule. Biofizika 10 no.2:349-351 '65. (MIRA 18:7)

1. Botanicheskiy institut imeni Komarova AN SSSR, Leningrad.

17(3)

AUTHORS:

Sapozhnikov, D. I., Eydel'man, Z. M., SOV/20-127-5-54/58  
Bazhanova, N. V., Popova, O. F.

TITLE:

The Inhibitory Effect of Hydroxylamine on the Light Reaction  
in the Course of Xanthophyll Transformation

PERIODICAL:

Doklady Akademii nauk SSSR, 1959, Vol 127, Nr 5, pp 1128-1131  
(USSR)

ABSTRACT:

In the most recent papers the participation of carotenoids in the transfer of oxygen in the course of the photosynthesis is assumed (Refs 1-5). The content of violaxanthine was reduced at illumination whereas that of lutein increased. This difference was reduced in the dark. Sapozhnikov Krasovskaya, and Mayevskaya (Ref 3) assumed an enzymatic nature of this mutual transformation of the two xanthophylls mentioned and the possible participation of this ferment system in the oxygen transfer. Furthermore it was proved that the violaxanthine formation was inhibited under anaerobic conditions (reaction in the dark) whereas the light reaction was not suppressed by the anaerobiosis. Since oxygen is transferred in the light reaction of the xanthophyll transformation it was important to investigate the inhibition conditions of this

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The Inhibitory Effect of Hydroxylamine on the Light  
Reaction in the Course of Xanthophyll Transformation

SOV/20-127-5-54/58

reaction. Hydroxylamine is a photosynthetic poison which acts as a specific inhibitor of the oxygen separation during the photosynthesis (Refs 6-9). Water weed (*Elodea canadensis*), i. e. the youngest shoot tips, 2 - 3 cm long, served as investigation object. After having been dried they were placed in boiling dishes with poison solutions of certain concentration. Figure 1 shows the results of a typical experimental series. A part of the boiling dishes with experimental- and control plants was exposed to the light of a 1000 watt lamp; the other one left in the dark. Various expositions (Fig 2) (2-120 minutes) in the poison solution and various poison concentrations (Fig 4) ( $1 \cdot 10^{-4}$  -  $6 \cdot 10^{-2}$  mol) as well as the illumination intensity (Fig 3) were tested. The following conclusions are drawn from the results: (1) The light reaction of the xanthophyll transformation may be completely inhibited by certain concentrations ( $4 \cdot 10^{-2}$  mol). (2) The concentration of the inhibitor necessary for the inhibition of the light reaction increases with rising light intensity. (3) The assumption concerning the enzymatic character of the light

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The Inhibitory Effect of Hydroxylamine on the Light  
Reaction in the Course of Xanthophyll Transformation

SOV/20-127-5-54/58

reaction of the mutual transformation of xanthophylls as well as concerning a close connection between this system and the oxygen transfer in the course of the photosynthesis is confirmed. There are 4 figures and 15 references, 6 of which are Soviet.

PRESENTED: April 23, 1959, by A. I. Oparin, Academician

SUBMITTED: March 16, 1959

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SAPOZHNIKOV, D.I.; EYDEL'MAN, Z.M.; BAZHANOVA, N.V.; MASLOVA, T.G.; POPOVA, O.F.;  
SHIRYAYEVA, G.A.

Characteristics of the light reaction of xanthophyll conversion under  
conditions of anaerobiosis. Bot.zhur. 49 no.10:1463-1465 0 '64.  
(MIRA 18:1)

1. Botanicheskiy institut imeni V.L.Komarova AN SSSR, Leningrad.

FOROVN

SOV 77-A-2-15/18

23(a) 23 (5)

AUTHOR: Lyalikov, K.S.

TITLE:

Successes of Soviet Electrophotography (Uspekhi sovetskoy elektrofotografii) A Scientific and Technical Conference on Questions of Electrophotography (Nauchno-tekhnicheskiye konferentsii po voprosam elektrofotografii)

PERIODICAL: Zhurnal nauchnoy i prikladnoy fotografii i kinematografii, 1959, Vol. 2, No. 2, pp 149-152 (USSR)

ABSTRACT:

This is an account of a scientific and technical conference on electrophotography, the first of its kind in the Soviet Union and evidently in the world. It was organized in Vil'nyus on December 29-30, 1958 by the Soviet Narodnogo Khozyaystva Litovskaya SSR (Council for National Economy of the Lithuanian SSR), the Gosudarstvennyy nauchnyy tsentr (State Scientific and Technical Committee of the USSR), the Ministry of the Lithuanian SSR and the Scientific Research Institute of Electrophotography. The conference, attended by over 300 scientific workers, was opened by the Deputy Chairman of the Council for National Economy of the Lithuanian SSR P.A. Kul'vets, after which the director of the Institute for Electrophotography, I.I. Zhilevich, reviewed the state and prospects for development of electrophotography in the USSR. He stated that research in this field should be carried out along the following lines: a) search for new photo-active materials with high dark resistance; b) physical research into internal photoeffect; c) development of photoreceptors with high sensitivity and development of the technology of the electrophotographic process, most of which he suggested also for O.S. Iopova. The sensitivity of electrophotographic layers in GOST units, N.S. Plavina (speaking also for I.I. Zhilevich, L.I. Krut'ko, M.M. Markovich, B.I. Kalinauskas and O.K. Suvayzida) reported on some research on the sensitization of a semiconductor in electrophotographic layers. V.M. Pridkin gave a report on highly sensitive electrophotographic layers and an electrophotocopying device, and reviewed the formation process of the latent electrophotographic image on the basis of the photo theory. He also described the design of an electrophotoreceptor for determining sensitivity by light and the circuit of a charge transfer device. V.M. Pridkin and the circuit of a final electrophotographic copying device. Finally, the final chapter describing the latent and then store on the mechanics and kinetics of the development of the latent electrophotographic image in liquid developers.

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SOV/77-A-2-15/18

Successes of Soviet Electrophotography, A Scientific and Technical Conference on Questions of Electrophotography

K.M. Vinogradov described some of the features of the electrostatic field of electrophotographic development. V.Yu. Kravchenko described the criterion of light sensitivity of the electrophotographic process. After the reports, a discussion took place on methods of determining the light sensitivity of electrophotographic layers. A.M. Chervashchev spoke on the prospects of developing polymeric processes using electric and magnetic forces. O.V. Gromov (speaking also for I.I. Zhilevich, A.A. Sukhly, V.A. Gordeyev, A.S. Puzha and Yu. I. Kovalytsa) reported on the development of electrophotographic reproducing equipment. A.S. Puzha (speaking also for I.I. Zhilevich, A.S. Borinovich, M.M. Galvudka and I.I. Rutkauska) reported on the use of electrophotographic methods in recording oscillographs and other recording instruments.

V.P. Iurchenko (speaking also for L.N. Galin) spoke on the possibility of electrophotographically recording images from electron-beam tubes. A.S. Korol (speaking also for M.A. Mitkha, V.I. Gorbunov, A.I. and K.I. Martynov) gave a detailed description of laboratory and machine methods of producing photosensitive layers (zinc oxide was used). A.A. Sukhly (speaking also for I.I. Zhilevich, O.V. Gromov, V.A. Gordeyev, K.V. Fedotov and T.M. Gelf) described a laboratory and industrial machine for producing photosensitive layers. T.A. Shishkina (speaking also for Ya.A. Gikman) reported on a method of examining electrophotographic material using an a/c bridge. S.I. Khotimovich (speaking also for A.I. Shkva and I.S. Chelkova) spoke on developing materials for electrophotography and ferrography, including developers giving a reverse image. S.I. Khotimovich reviewed methods of measuring the electrostatic potentials of electrophotographic layers and the influence of various electrode materials on the electrophotographic process. Until as this causes self-discharge. V.V. Kravtsov (speaking also for A.S. Gordeyev, V.I. Gorbunov and Ya. S. Kuyfets) spoke on the practice of producing velvet papers in an electrostatic field, and showed samples produced by the Irigishkva paper factory. Ye.I. Melnikovskiy then gave a historical review of the development of electrophotographic methods in which he paid tribute to the work of the Scientific Research Institute of Electrophotography in Vil'nyus and the Institut Poligraficheskogo Mashinostroyeniya (Moscow) (polygraphic machine-building institute (Moscow)). Debates were then held

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on methods of measuring the potential of charged electro-  
photographic layers, the vibration pick-up most-used  
was shown in A.I. Ikonov's report to be not always  
accurate. I.G. Gerasimov stated that the bad influence  
of the electrode on the electrostatic field is estimated if the  
electrode is connected to ground. The electrostatic field  
up is connected to it by a shield cable. In the pick-  
up the research of Academician A.M. Terenin and Ye.M.  
Putyeko should be considered as the basis of all work  
on electrophotographic papers with ZnO, as they were  
the first to show the possibility of optical sensiti-  
vation of the internal photoeffect in ZnO. M.M. Gol-  
vidis then gave a report on the depositing of charges  
by a corona discharge. A.I. Kharlamov and A.P.  
Lengua reviewed some of the results of the use of  
electrographic methods in radiography. L.I. Myunko  
(speaking also for I.I. Zhilovich, I.N. Pavlov, Yu.K.  
Vishchak and Yu.A. Zibutis) reported on relaxation pro-  
cesses in semiconductor layers, using a vibration electro-  
meter. Yu.K. Vishchak gave a report on research on some  
physical properties of the polycrystalline layers of  
selenium cadmium. I.P. Likhtyavichyus spoke on some  
of the photoelectric properties of ZnO and ZnS. S.M.  
Kagan reported on methods of obtaining medium  
light-sensitive layers including sublimation and ther-  
mal treatment; it was also found that the sensitivity  
of the layers increased after storage for 1.5 to 2 months  
at room temperature. P.M. Podvigalshin (speaking also  
for S.G. Gerasimov) spoke on research into the elec-  
trical properties of electrophotographic layers of  
amorphous selenium and powdered zinc oxide. M.K.  
Shiklov (speaking also for A.S. Turyaytis) discussed  
the production of selenium layers and some of their  
properties. Finally the following reports on ferro-  
magnetography were delivered: 1) B.Ya. Karmachev,  
V.I. Zolotarev, "Electrodeposition of Magneto-Hard Alloys  
with Given Magnetic Characteristics" 2) M.M. Arutyunov,  
"Visualization of Magnetic Oscillations by the Ferro-  
graphic Method" 3) A.S. Turyaytis, "Ferrographic Recording  
of Magnetic Fields" 4) I.I. Zhilovich, I.N. Pavlov, R.  
Ye. Bushchik, "Ferrographic Recording of Magnetic Fields  
in Non-Pressure Regions" 5) I.I. Zhilovich, I.N. Pavlov,  
also an exhibition showing the work of the Electro-  
graphic Institute. The most important conclusion of  
the conference was that a solid approach had been made  
to the possibility of wide technical use of the methods  
of electrography. It was considered that although work  
in this field actually started only in 1955-56 it has covered as much ground  
as the USA in 10 years. This admitting that it was  
easier to reproduce results already achieved than to be  
the first to arrive at them, the conference observed  
that the Americans took good care that no important  
information appeared in the literature available.

Card 10/10

POPOVA, O. I.

POPOVA, O. I.: "A study of the colored complexes of titanium and their use in colorimetric analysis." Min Higher Education USSR. Kiev State U imeni T. G. Shevchenko. Kiev 1956  
(Dissertation for the degree of Candidate in Chemical Science)

SO: Knizhnaya Letopis', No36, 1956, Moscow.

BABKO, A.K.; POPOVA, O.I.

~~Study of titanium complex compounds with chromotropic acid and~~  
1,8-dioxynaphthalene. Zhur. neorg. khim. 2 no.1:138-146 Ja '57.  
(MLBA 10:4)

1. Kiyevskiy gosudarstvennyy universitet.  
(Titanium compounds) (Complex compounds) (Naphthalenedisul-  
fonic acid)

Popova, O. I.

✓6800\* Russian Study of Titanium-Pyrocatechin Complexes.  
Izucheniye pirokatekhinatnykh kompleksov titana. A. K.  
Babko and O. I. Popova. Zhurnal Neorganicheskoi Khimii. 2.  
Jan. 1957. 2-44. 188.  
Optical characteristics of pyrocatechin Ti complexes. Effect of  
pH on the formation and effect of concentration. Ionization  
series equilibrium.

mm

POPOVA, O. I.

USSR/Cultivated Plants. Fodder Plants.

M

Abs Jour : Ref Zhur-Biol., No 15, 1956, 68240

Author : Popova, O. I.  
Inst : Onsk Agricultural Institute.  
Title : A Comparative Study of Annual Grasses.

Orig Pub : Tr. Onskogo s.-kh. in-ta, 1957, 22, No 1,  
149-159

Abstract : In 1953-1954, experiments at the Onsk Agricultural Institute have shown that the highest hay yields were given by broomcorn (66.8 centners/hectare), grain sorgo (55.7 centners/hectare), Buchan chumiza (64.1 centners/hectare), millet (29.1 centners/hectare), and nohar (28.2 centners/hectare). Sudan grass, nohar, and millet gave lower hay yields in 1954, and oats in

Card : 1/2

AUTHOR: Babko, A. K., and Popova, O. I.

73-3-16/24

TITLE: Determination of Titanium in Steels by Extracting its Complex with 1,8- $\alpha$ -Dihydroxynaphthalene. (Opredeleniye Titana v Stalyakh Ekstragirovaniyem Yego Kompleksa s 1,8-Dioksinaftalinom)

PERIODICAL: Ukrainskiy Khimicheskii Zhurnal, 1957, Vol. 23, No.3, pp. 376-380 (USSR).

ABSTRACT: 1,8-dihydroxynaphthalene was shown to form complexes with titanium which has analogous properties as titanium-chromotropic complexes but can be extracted by means of organic solvents (Ref. 6). 1,8-dihydroxynaphthalene forms 2 compounds with titanium: a red complex at pH 1 - 4 ( $\lambda_{\max} = 490 \text{ m}\mu$ ) and a yellow complex at pH  $\geq 6$  ( $\lambda = 430 \text{ m}\mu$ ).

These complexes are insoluble in water, form colloidal solutions which coagulate after 2 days. Chemico-physical analyses showed that the relation titanium: dihydroxynaphthalene = 1:2 in the red complex and 1:3 in the yellow complex, the first complex being formed in a more acid medium, the second when the pH  $\geq 6$ . The dependence of the formation of the complexes on the pH, the effect of some anions as well as the properties of the complexes were

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73-3-16/24

Determination of Titanium in Steels by Extracting its Complex  
with 1,8- $\alpha$ -Dihydroxynaphthalene.

investigated. The complex can be extracted with butanol, ethyl ether, isobutanol and benzene at pH 2 - 8. One extraction only is required. Figure 1 shows the effect of excess 1,8-dihydroxynaphthalene ( $H_2R$ ) during the extraction of the complex at pH 4 and pH 6. The intensity of coloration of the extract of the complex is proportional to the titanium content in the solution (Fig. 2). Measurements were carried out on a photometer  $\phi M$  at an effective wavelength 465  $m\mu$ . (pH = 4).  $Co^{2+}$ ,  $Ni^{2+}$  and  $Cu^{2+}$  ions do not impede the extraction of the complex. Fe(III) forms a green precipitate with 1,8-dihydroxynaphthalene, but when converted into divalent Fe it will not react with the diol. Ascorbic acid, thiosulphate, Na-sulphite, hydroxylamine and hydrazine sulphate were used as reducing agents. Figures 3 and 4 give the light absorption curves of titanium-diol complexes in aqueous solution at pH 2 - 7 and in butanol at the same pH. Measurements were carried out on a spectrophotometer  $C\phi$  - 4. Optimum conditions for the red complex are at  $\lambda \sim 540 m\mu$  and for the yellow complex at  $\lambda \sim 430 m\mu$ . A method for the determination of titanium in Card 2/3 steels by extraction without separating the chrome is given.

5(2)

AUTHOR:

Popova, O. I.

SOV/32-25-2-11/78

TITLE:

The Determination of Small Ti-Quantities in Tin Alloys  
(Opredeleniye malykh kolichestv titana v olovyannykh splavakh)

PERIODICAL:

Zavodskaya Laboratoriya, 1959, Vol 25, Nr 2, p 148 (USSR)

ABSTRACT:

It has been observed that in the determination of Ti in tin alloys by chromotropic acid (I) in concentrated sulfuric acid (Ref 1) on certain conditions tin does not react with (I). The Ti-determination is not disturbed by the presence of niobium and tantalum. A portion of 0.5 to 1.0 g is heated and dissolved in concentrated sulfuric acid, a 1 % freshly prepared solution of (I) is added to the solution and filled up with concentrated sulfuric acid to a certain volume. The color intensity obtained is compared with a scale of standard solutions. A table gives the results of the analyses of artificial mixtures containing 1 g of zinc each. In the analysis of an alloy by the method described an average of 0.00175 % Ti was found, while the result obtained by the addition method was 0.0017 %. There are 1 table and 2 references.

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The Determination of Small Ti-Quantities in Tin  
'Alloys

SOV/32-25-2-11/78

ASSOCIATION: Institut metallokeramiki i spetsial'nykh splavov Akademii  
nauk USSR (Institute of Powder Metallurgy and Special  
Alloys of the Academy of Sciences, UkrSSR)

Card 2/2

POPOVA, O.I., KABANNIK, G.T.

Acid resistance and methods of analysis of titanium, zirconium, niobium and tantalum nitrides. Zhur. neorg. khim. 5 no.4:930-934 Ap '60. (MIRA 13:7)

1. Institut metallokeramiki i spetsial'nykh splavov AN USSR.  
(Titanium nitride) (Zirconium nitride)  
(Niobium nitride) (Tantalum nitride)

S/700/61/000/006/007/016  
D217/D304

AUTHORS: Popova, O. I. and Kabannik, G. T.

TITLE: Chemical properties and analysis of some nitrides

SOURCE: Akademiya nauk Ukrainskoy SSR. Institut metallokeramiki i spetsial'nykh splavov. Seminar po zharostoykim materialam. Kiyev, 1960. Trudy no. 6: Khimicheskiye svoystva i metody analiza tugoplavkikh soedineniy. Kiyev, Izd-vo AS UkrSSR, 1961, 64-68

TEXT: A few chemical properties and methods of analyzing the following nitrides are discussed: BN,  $\text{Si}_3\text{N}_4$ , TiN, ZrN, NbN, TaN,  $\text{Cr}_2\text{N}$ , CrN, VN,  $\text{V}_2\text{N}$ , AlN, and MgN. These nitrides, except AlN and MgN, are resistant to various chemical reagents. All nitrides decompose on fusion with alkalis and  $\text{Na}_2\text{O}_2$ . MgN and AlN are less stable. MgN decomposes in air (it must be kept in ampoules) and is easily dissolved in acids; AlN dissolves in dilute alkalis.  $\text{V}_2\text{N}$  and  $\text{Cr}_2\text{N}$  dis-

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Chemical properties and ...

S/700/61/000/006/007/018  
D217/D304

solve in some acids, whereas VN and CrN dissolve only in nitric and perchloric acid, respectively. The solubility of nitrides of Ti, Zr, Nb, Ta and Cr in various acids, acid mixtures and alkaline solutions was studied, and it was established that TaN and CrN are the most, and ZrN and Cr<sub>2</sub>N the least, resistant to the action of various solvents. The results of this study of the solubility of the nitrides were applied to the decomposition of the specimens for analysis. Methods for analyzing some of the nitrides are given. There are 4 tables and 14 references: 7 Soviet-bloc and 7 non-Soviet-bloc.

ASSOCIATION: Institut metallokeramiki i spetsial'nykh splavov AN USSR (Institute of Powder Metallurgy and Special Alloys AS UkrSSR)

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35054

S/700/61/000/006/009/018  
D267/D304

18.1200

AUTHORS: Samsonov, G. V., Vereykina, L. L. and Popova, O. L.

TITLE: Investigating chemical stability and methods of chemical analysis of Ti-P and Cr-P alloys

SOURCE: Akademiya nauk Ukrainskoy SSR. Institut metallokeramiki i spetsial'nakh splavov. Seminar po zharostoykim materialam. Kiyev, 1960. Trudy no. 6: Khimicheskiye svoystva i metody analiza tugoplavkikh soyedineniy. Kiyev, Izd-vo AS UkrSSR, 1961, 75-79

TEXT: The monophosphides (TiP and CrP) were prepared by passing  $\text{PH}_3$  over heated metal powder under O-free argon. The phosphine was obtained by the acid decomposition of AlP. To obtain TiP it is recommended carrying out two 6-hour phosphidizations at  $1000^\circ\text{C}$ , and for obtaining CrP -- a single 7-hour phosphidization at  $850^\circ\text{C}$ . The reactions proceed faster when metal hydrides are substituted for the metals. After 10 - 12 hours' boiling,  $\text{TiP}_{0.96}$  was found to be

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Investigating chemical stability ...

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D267/D304

remained undissolved in the presence of strong oxidants. Therefore, when analyzing CrP, the sample is dissolved by heating in  $\text{HNO}_3$  (conc.) +  $\text{H}_2\text{SO}_4$  (conc.) or  $\text{H}_2\text{SO}_4$  (conc.) +  $\text{NH}_4\text{S}_2\text{O}_8$ , after which the excess of oxidant is removed and the total Cr content is determined volumetrically. Combined P is determined gravimetrically by precipitating with magnesia mixture. Free Cr in the phosphide is determined by making use of the fact that  $\text{H}_2\text{SO}_4$  (1:4) dissolves free Cr, but does not dissolve CrP. The volumetric method is then used for determining Cr in the solution. There are 4 tables and 13 references: 4 Soviet-bloc and 9 non-Soviet-bloc. The reference to the English-language publication reads as follows: I. Haugton, Iron Steel Inst. (London), 115, 417 (1927).

ASSOCIATION: Institut metallokeramiki i spetsial-nykh splavov AN USSR (Institute of Powder Metallurgy and Special Alloys AS UkrSSR)

Card 3/3

POPOVA, O.I.; KORNILOVA, V.I.

Analysis of high-melting alloys containing zirconium (titanium) and tungsten. Zhur.anal.khim. 16 no.5:651-652 S-O '61. (MIRA 14:9)

1. Institute of Metalloceramics and Special Alloys, Academy of Sciences, Ukrainian S.S.R., Kiyev.

(Zirconium-tungsten alloys)

PCPOV., C.I.

Служба: Уполном. Упр. Мин. Упр. № 10-4/10.

(1-24)

1. Институт металлообработки и сварочный институт.  
(Черноморский завод)



ACCESSION NR: AP4043462

S/0075/64/019/008/0980/0984

AUTHORS: Nazarchuk, T.N.; Popova, O.I.; Kugay, L.N.; Dzerzhanovskaya, Ye.V.; Kabannik, G.T.; Boremskaya, S.F.; Chugunnaya, N.K.

TITLE: Analysis of rare earth alloys with certain metals and oxides

SOURCE: Zhurnal analiticheskoy khimii, v. 19, no. 8, 1964, 980-984

TOPIC TAGS: complexometric titration, rare earth analysis, lanthanum oxide, magnesium oxide, scandium oxide, yttrium oxide, chromium oxide, nickel oxide, aluminum oxide

ABSTRACT: Two methods of separation and determination of rare earth elements were worked out. The first method involved titration with complexon III at different pH of the solution in the presence of different indicators. Here the fact that tetravalent elements such as titanium and zirconium, form complexes in strongly acid solutions (pH = 1), trivalent metals at pH 2 - 3 and alkaline earth elements at pH 10 - 11 was taken into account. The second method involved the use of masking substances such as potassium cyanide, triethanolamine, ammonium fluoride, thyron, 2,3-dimercapto-propanol. The analysis

Card 1/2

POPOVA, O.I.; GODOVANNAYA, I.N.

Complexometric analysis of some binary alloys. Zhur. anal. khim.  
20 no.3:355-358 '65. (MIRA 18:5)

1. Institut problem materialovedeniya AN UkrSSR, Kiyev.

SAMSONOV, G.V.; PILIPENKO, A.T., prof., doktor khim. nauk; NAZARCHUK, T.N., kand. khim. nauk; Primalni uchastiye: POPOVA, O.I., kand. khim. nauk; KOSOLAPOVA, T.Ya.; OBOLONCHIK, V.A.; KOTLYAR, G.Kh., mladshiy nauchnyy sotr.; KUCHAY, L.N.; KOPYLOVA, V.P.; KABANNIK, G.T.; KLIBUS, A.Kh.; MODYLEVSKAYA, K.D.; RADZIKOVSKAYA, S.V.; NIKITINA, Ye.A., red.; KAMAYEVA, O.M., red. izd-va; KARASEV, A.I., tekhn. red.

[Analysis of high-melting compounds] Analiz tugoplavkikh soedinenii. Moskva, Metallurgizdat, 1962. 256 p. (MIRA 15:7)

1. Chlen-korrespondent Akademii nauk USSR (for Samsonov).  
(Intermetallic compounds—Analysis)  
(Nonmetallic materials—Analysis)

S/081/62/000/019/016/053  
B144/B180AUTHORS: Popova, O. I., Kabannik, G. T.

TITLE: Chemical properties and analysis of certain nitrides

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 19, 1962, 121, abstract.  
19D111 (Byul. In-t metallokeram. i. spets. splavov AN USSR,  
no. 6, 1961, 64 - 68)

TEXT: The solubility of Ti, Zr, Nb, Ta, and Cr nitrides was studied in different acids ( $\text{HCl}$ ,  $\text{H}_2\text{SO}_4$ ,  $\text{HNO}_3$ ,  $\text{HClO}_4$ , etc.), mixtures thereof, and in alkaline solutions ( $\text{NaOH}$ ,  $\text{NaOH} + \text{H}_2\text{O}_2$ , etc.). The nitrides were obtained by nitriding the metals at high temperatures..  $\text{TaN}$  and  $\text{CrN}$  were the most resistant,  $\text{Cr}_2\text{N}$  the least. Nitrogen in the nitrides is determined by the Kjeldal method after dissolving the sample in concentrated  $\text{H}_2\text{SO}_4$  ( $\text{ZrN}$ ,  $\text{NbN}$ ) in a  $\text{H}_2\text{SO}_4 - \text{K}_2\text{SO}_4$  mixture ( $\text{TiN}$ ,  $\text{TaN}$ ,  $\text{BN}$ ,  $\text{VN}$ ), in  $\text{H}_2\text{SO}_4$  (1 : 4) ( $\text{Cr}_2\text{N}$ ,  $\text{FeN}$ ), or in 40%  $\text{NaOH}$  solution ( $\text{AlN}$ ).  $\text{CrN}$  and  $\text{Si}_3\text{N}_4$  are fused with a  $\text{PbO}_2 + \text{PbCrO}_4$

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S/081/62/000/019/012/053  
B144/B180

AUTHORS: Samsonov, G. V., Vereykina, L. L., Popova, O. I.

TITLE: Methods of chemical analysis for, titanium - phosphorus and chromium - phosphorus alloys, and a study of their resistance to chemical corrosion

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 19, 1962, 119, abstract 19D103 (Byul. In-t metallokeram. i spets. splavov AN USSR, no. 6, 1961, 75 - 79)

TEXT: The resistance of Cr and Ti monophosphides obtained by passing phosphine over the heated metal powder to chemical corrosion was studied in acid ( $H_2SO_4$ ,  $HNO_3$ ,  $HCl$ ,  $HF$ , and mixtures of them) and alkaline ( $NaOH + H_2O_2$ ,  $NaOH +$  bromine water) media and methods of analysis developed. For analyzing Ti monophosphide, the sample (0.1 - 0.15 g) is dissolved in a mixture of 40%  $HF$  and concentrated  $HNO_3$  in a Pt dish. The solution is slightly evaporated, 30 ml of 35% tartaric acid solution and a small quantity of solid  $H_3BO_3$  are added and the mixture is diluted to 200 ml. To 10 - 25 ml  
Card 1/3

S/081/62/000/019/012/053

B144/B180

## Methods of chemical analysis ...

of the solution obtained 10 ml concentrated  $\text{HNO}_3$  and 15 mg  $\text{NH}_4\text{NO}_3$  are added, it is heated to  $60^\circ\text{C}$ , P is precipitated by adding 100-ml molybdate solution and it is left for one night. Then it is filtered through a fine filter, the precipitation is thoroughly washed and transferred together with the filter into the NaOH titrant whose excess is backtitrated with phenolphthalein as indicator. The Ti content is determined from a weighed portion separated by precipitation with cupferron from the sulfate solution or titrimetrically after reduction to  $\text{Ti}^{3+}$ . For the analysis of Cr monophosphide, the sample (0.1 - 0.15 g) is dissolved by heating in a mixture of concentrated  $\text{H}_2\text{SO}_4$  and  $\text{HNO}_3$  or in a mixture of  $\text{H}_2\text{SO}_4$  and  $(\text{NH}_4)_2\text{S}_2\text{O}_8$ . The solution is evaporated till evolution of a white fume and after cooling and dilution its Cr content is determined by oxidation to  $\text{Cr}^{6+}$  with  $(\text{NH}_4)_2\text{S}_2\text{O}_8$  (catalyst  $\text{AgNO}_3$ ) and titrating  $\text{Cr}^{6+}$  with 0.1 N Mohr's salt solution (indicator phenyl anthranilic acid). The P content is determined gravimetrically by precipitation with magnesia mixture. In order to determine the free Cr in Cr monophosphide, 0.2 - 0.25 g of the latter is treated by heating with  $\text{H}_2\text{SO}_4$  (1 : 4) (in this case only the free Cr passes into the solution); the

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Card 2/3

S/137/62/CCC/CC8/C59/C55  
A006/A101

AUTHORS: Popova, O. I., Kabannik, G. T.

TITLE: Chemical properties and analysis of some nitrides

PERIODICAL: Referativnyy zhurnal, Metallurgiya, no. 8, 1962, 7, abstract 8K44  
("Byul. In-t metallokeram. i spets. splavov AN UkrSSR", 1961, no. 6,  
64 - 68)

TEXT: The authors investigated the solubility of Ti, Zr, Nb, Ta and Cr nitrides in various acids, their mixtures and alkaline solutions. It is shown that maximum stability against the effect of different solvents is offered by Ta and Cr nitrides ( $\text{CrN}$ ), and least stability by Zr and Cr nitrides ( $\text{Cr}_2\text{N}$ ). Methods of analyzing some nitrides are given. For instance, when analyzing Si nitride, the batch is alloyed with  $\text{Na}_2\text{O}_2$ . The cold melt is lixiviated with water and further analysis is conducted with the use of the hydrochloride method. The determination of free Si is based on its solubility in 1% NaOH solution. The Si nitride is then not dissolved. In the solution obtained the Si content is determined by the colorimetric method from the yellow color of the molybdenum silicic acid.

[Abstracter's note: Complete translation]  
Card 1/1

L. Vorob'yeva

POPOVA, L. I.

PHASE I BOOK EXPLOITATION

SOV/5994

Akademiya nauk Ukrainiskoy SSR. Institut metallokeramiki i spetsial'nykh splavov. Seminar po zharostoykim materialam. Kiyev, 1960.

Trudy Seminara po zharostoykim materialam, 19-21 aprelya 1960 g. Byulleten' no. 6: Khimicheskiye svoystva i metody analiza tugoplavkikh soedineniy (Transactions of the Seminar on Heat-Resistant Materials of the Institute of Powder Metallurgy and Special Alloys of the Academy of Sciences of the Ukrainian SSR. Held 19-21 April, 1960. Bulletin no. 6: Chemical Properties and Methods of Refractory Compound Analysis). Kiyev, Izd-vo AN UkrSSR, 1961. 124 p. 1500 copies printed.

Sponsoring Agency: Akademiya nauk Ukrainiskoy SSR. Institut metallokeramiki i spetsial'nykh splavov.

Editorial Board: I. N. Frantsevich; G. V. Samsonov, Resp. Ed.; I. M. Fedorchenko, V. N. Yermenko, V. V. Grigor'yeva, and T. N. Nazarchuk; Tech. Ed.: A. A. Matveychuk.

Card 1/5



Transactions of the Seminar (Cont.)

SOV/5994

**PURPOSE:** This collection of articles is intended for chemists, engineers, workers at scientific research institutes and plant laboratories, senior students, and aspirants at chemical and metallurgical schools of higher education.

**COVERAGE:** Articles of the collection present the results of studies of the chemical properties of refractory compounds (carbides, borides, nitrides, phosphorides, silicides), refractory and rare metals, and their alloys, and some original methods of analyzing these materials, which are now being utilized in the new fields of engineering. No personalities are mentioned. Each article is accompanied by references, mostly Soviet.

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Samsonov, G. V. Refractory Compounds, Their Properties, Pro-  
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POPOVA, O. I.

PHASE I BOOK EXPLOITATION

SOV/6030

Samsonov, G. V., Corresponding Member, Academy of Sciences UkrSSR; A. T. Pilipenko, Doctor of Chemical Sciences, Professor; T. N. Nazarchuk, Candidate of Chemical Sciences; O. I. Popova, Candidate of Chemical Sciences; and T. Ya. Kosolapova, V. A. Obolonchik, G. Kh. Kotlyar, L. N. Kuchay, V. P. Kopylova, G. T. Kabanik, A. Kh. Klibus, K. D. Modylevskaya, and S. V. Radzikovskaya.

Analiz tugoplavkikh soyedineniy (Analysis of Refractory Compounds) Moscow, Metallurgizdat, 1962. 256 p. 3250 copies printed.

Ed.: Ye. A. Nikitina; Ed. of Publishing House: O. M. Kamayeva; Tech. Ed.: A. I. Karasev.

PURPOSE: This book is intended as a laboratory manual for personnel in plant laboratories of the machinery, chemical, and aircraft industries and scientific research institutes. It can also be used by chemistry students at universities and schools of higher education.

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Analysis of Refractory (Cont.)

SOV/6030

**COVERAGE:** The book contains data from the literature and from laboratory research on the chemical and mechanical properties, crystalline structure, chemical analysis, production, and industrial and other applications of silicon carbide and other refractory compounds. Methods of determining the basic components of refractory compounds (carbon, boron, nitrogen, and silicon) are reviewed and detailed methods for the chemical analysis of all presently known refractory compounds given. The authors are associated with the Institut metallokeramiki i spetsial'nykh splavov, AN SSSR (Institute of Powder Metallurgy and Special Alloys, Academy of Sciences USSR). No personalities are mentioned. There are 327 references: 175 Soviet and the remainder mainly English and German.

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AVAILABLE: Library of Congress

SUBJECT: Metals and Metallurgy

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BN/pw/bmc  
10-30-62

POPOVA, O.I.; KABANNIK, G.T.

Chemical properties and analysis of some nitrides. Biul. Inst.  
metallor. i spets. splav. AN USSR no.6:64-68 '61. (MIRA 15:2)

1. Institut metallokeramiki i spetsial'nykh splavov AN USSR.  
(Nitrides)

HAZARCHUK, Tamara Nikolayevna; LITKOVA, Oksana Ivanovna, SANCHEV,  
G.V., ed.; BOGOMOLOVA, L.N., ed.; FOMIN, N.N.,  
ed.

[Complexometric analysis of ceramic metal and ceramic  
materials and of certain alloys] Kompleksometricheskii  
analiz metallokeramicheskikh i keramicheskikh materialov  
i nekotorykh spлавov. Kiev, Naukov dumka, 1965. 120 p.  
(RISA 18:9)

1. Chlen-korrespondent AN Ukr.SSR (for Sanchov).



POPOVA, O. L.

USSR/Microbiology. Antibiosis, and Symbiosis,  
Antibiotics.

F-2

Abs Jour : Ref. Zhur-Biologiya, No 1, 1957, 515  
Author : S. F. Gauze, O. L. Popova, G. V. Kochetkova  
Inst :  
Title : New Method of Selection of the Producer  
of Albomycin  
Orig Pub : Antibiotiki, 1956, 1, No 1, 18-20  
Abstract : When a suspension of spores of Actinomyces  
subtropicus, the producer of albomycin,  
is subjected to ultra-violet light, in  
the subsequent selection it was not  
possible to isolate strains with a greater  
productivity of albomycin (1) than those  
isolated from the initial culture. No  
results were obtained also in the attempt  
to derive a more active variant by

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USSR/Microbiology. Antibiosis, and Symbiosis,  
Antibiotics.

F-2

Abs Jour : Ref. Zhur-Biologiya, No 1, 1957, 515

Abstract : colony has grown from 1,000, 20,000 and 40,000 spores respectively. Streptomycin resistant variants which freely develop in 150 gamma/ml varied considerably in their morphological and physiological properties. A change in the color of the mycelium was observed in 15 cases out of 200. In a small number of strains of 524 streptomycin resistant forms the formation of l. exceeded by 150 to 200 percent the formation of l. from the initial culture, and this index was maintained by a number of generations.

Card 3/3



GAUSE, G.F.; KOCHETKOVA, G.V.; PREOBRAZHENSKAYA, T.P.; KUDRINA, E.S.;  
SVESHENIKOVA, M.A.; POPOVA, G.I.

The use of actinophages in the search for antiviral antibiotics.  
J. Hyg. Epidemiol., Praha 1 no.1:63-69 1957.

1. Institute for Antibiotics Research of the Academy of Medical Sciences  
of the U.S.S.R., Moscow.

(ACTINOMYCES,

actinophages, in research on antiviral antibiotics)

(ANTIBIOTICS,

antiviral, use of actinophages in research)

(BACTERIOPHAGE,

actinophage in research on antiviral antibiotics)

GAUZE, G.F.; KOCHETKOVA, G.V.; PREOBRAZHenskAYA, T.P.; KUDRINA, Ye.S.;  
SVESHNIKOVA, M.A.; POPOVA, O.L.

Study of the inhibiting effect of actinomycetes on actinophages  
[with summary in English]. Mikrobiologiya 26 no.6:729-735 H-D '57.  
(MIRA 11:3)

1. Institut po izyskaniyu novykh antibiotikov AMN SSSR, Moskva.  
(MICROORGANISMS,  
actinomycetes, inhib. eff. on actinophages (Rus)

SAMSONOV, G.V.; VEREYKINA, L.L.; POPOVA, O.L.

Study of the chemical stability and methods of chemical analysis  
of titanium-phosphorous alloys and chromium-phosphorous alloys.  
Biul.Inst.metaloker. i spets. splav. AN URSR no.6:75-79 '61.  
(MIRA 15:2)

1. Institut metallokeramiki i spetsial'nykh splavov AN USSR.  
(Titanium-phosphorous alloys)(Chromium-phosphorous alloys)

GAUZE, G.F.; MAKSIMOVA, T.S.; POPOVA, O.L.; BRAZHNIKOVA, M.G.; USPENSKAYA, T.A.;  
ROSSOLIMO, O.K.

Mutomycin, a new antibiotic produced by *Actinomyces atroolivaceus*.  
Antibiotiki 4 no.3:20-23 My-Je '59. (MIRA 12:9)

1. Institut po izyskaniyu novykh antibiotikov AMN SSSR.  
(ANTIBIOTICS,

mutomycin, prod. by *Actinomyces atroolivaceus*  
& pharmacol. (Rus))

KOCHETKOVA, G.V.; POPOVA, O.L.; BOBKOVA, T.S.; TOROPOVA, Ye.G.

Inactivating effect of some new antibiotics produced by  
Actinomyces on actinophages in vitro and in vivo. Antibiotiki  
3 no.5:17-21 S-0 '58. (MIRA 12:11)

1. Laboratoriya vydeleniya i kul'tivirovaniya produktentov (zav. -  
prof.G.F.Gauze) Instituta po izyskaniyu novykh antibiotikov AMN  
SSSR.

(BACTERIOPHAGE,  
actinophage, inactivation by antibiotics prod.  
by Actinomyces (Rus))  
(ACTINOMYCES,  
same)  
(ANTIBIOTICS,  
Actinomyces-prod., inactivation of actinophage  
(Rus))



Popova, O.L.

1.  
Virus/Virology - Bacterial Viruses (Phages)

Abstr Jour : Ref Zhur - Biol., No 19, 1958, 85765  
 Author : Gause, G.F., Kochetkova, G.V., Prokhorovskaya, G.P.,  
 Gurina, Ye.G., Sveshnikova, N.A., Popova, O.L.  
 Inst :  
 Title : Studies of the Suppressive Effects of Actinomycetes on  
 Actinophages.  
 Orig Pub : Mikrobiologiya, 1957, 26, No 6, 729-735

Abstract : Of 9 actinophages isolated from the soil only 2 were dis-  
 tinguished by specificity of action, while the others  
 were polyvalent. Comparative studies of the actinophage  
 and antibacterial activity of 1000 strains of Actinomyces  
 strains showed that of 546 strains which suppressed bacte-  
 rial growth, 311 also suppressed actinophages. In the  
 titration of actinophages on cultures of Actinomyces  
 which did not suppress bacterial growth, 247 of 546 strains  
 of actinophages. Of 576 cultures of Actinomyces with

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actinophage activity, 279 (48%) acted against 1, 147 (24%)  
 acted against 2, 66 (11%) acted against 3, 4 (7%) acted  
 against 4, 21 (4%) acted against 5, and 1 (0.2%) acted  
 against 6 different phages. -- G.V. Naumovskaya

Card 2/2

USSR/Virology - Bacterial Viruses (Phages).

Abstr Jour : Ref Zhur Biol., No 6, 1959, 23781  
 Author : Gause, O.F., Kochetkova, O.V., Fedotrubushchaya, T.P.,  
 Kudrina, Ye. S., Sveshnikova, M.A., Popova, G.L.  
 Inst :   
 Title : Actinophages as Test-Objects in a Search for Anti-Virus  
 Antibiotics.  
 Orig Pub : Zh. obshchey, i spetsial., mikrobiol. i immunit., 1957,  
 1, No 1, 53-58

Abstract : The ability was studied of 1000 cultures of Actinophages,  
 isolated from soils of various geographic locations, to  
 suppress four cultures of bacteria and six various Actino-  
 phages, of which four were Polypages. It was found  
 that about one-half of the tested Actinophages are  
 able to suppress one or several Actinophages in the ex-  
 periment. Actinophages were suppressed by Actinophages  
 with antibacterial activity as well as by Actinophages

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which did not possess antibacterial activity. It was  
 noted that Actinophages able to suppress a combination  
 of 4 Actinophages (No 2671, 2761, 250, and 3087) were  
 found most frequently; these Actinophages turn out to  
 be most convenient test-object in a selection of Actino-  
 phages of cultures which produce antiviral antibiotics.  
 -- Th.I. Naumenko

Card 2/2

CHERVONSKIY, V.L.; POPOVA, O.M.

Antigen for the Miyagawanella complement fixation reaction. Vop.  
virus 4 no.1:68-71 Ja-F '59. (MIRA 12:4)

1. Institut virusologii AMN SSSR, Moskva.  
(ORNITHOSIS, immunol.  
complement fixation, antigen (Rus))  
(COMPLEMENT FIXATION  
fixation reaction in ornithosis, antigen (Rus))

TERSKIKH, I.I.; ~~POPOVA, O.M.~~

Emergency prophylaxis for ornithosis by means of dibiomycin;  
development of a scheme for its use. Vop.virus. 7 no.6:712-715  
N-D '62. (MIRA 16:4)

1. Institut virusologii imeni D.I.Ivanovskogo AMN SSSR, Moskva.  
(ORNITHOSIS) (DIBIOMYCIN)

ACC NR: AP602155

(V)

SOURCE CODE: JR/0412/66/000/003/0312

AUTHOR: Nikolov, Z. V.; Popova, O. M.

ORG: Department of Epidemiology, Institute for Specialization and Postgraduate Training of Physicians (Kafedra epidemiologii Instituta spetsializatsii i usovershenstvovaniya vrachey, Sofia); Virology Institute, Academy of Medical Sciences, SSSR (Institut virusologii im. D. I. Ivanovskogo AMN SSSR)

TITLE: Effect of chemical disinfectants on ornithosis virus

SOURCE: Voprosy virusologii, no. 3, 1966, 375

TOPIC TAGS: disinfectant, virology, ornithosis, ornithosis virus, *VIRUS*

ABSTRACT:

The following disinfectants destroy ornithosis viruses and are recommended for use at sites of infection: a 5—10% solution of iodine chloride will inactivate all viruses in 1 hr; 3—5% hydrogen peroxide does the same in 3 hr; 0.1—0.2% dichlorohydantoin inactivates viruses in 1 hr, and a 0.5% solution of benzylchlorophenol requires 3 hr.

[W.A. 50; CBE No. 10]

SUB CODE: 06/ SUBM DATE: none/

Cord 1/1

POPOVA, G.M.; BEREZINA, O.N.

Effect of previous X-ray irradiation on the susceptibility of mice to infection by ornithosis virus aerosol. Vopr. virus. i bakt. 21:6. Moscow '64.

1. Institut virusologii imeni Frankovskogo AMN SSSR, Moscow.

VAVILOV, P.P.; VERKHOVSKAYA, I.N.; KODANEVA, R.P.; POPOVA, O.N.

Growth and development of *Vicia faba* L. under the conditions of increased content of U and Ra. Radiobiologiya 3 no.1:132-138 '63. (MIRA 16:2)

1. Laboratoriya radiobiologii Komi filiala AN SSSR, Syktyvkar.  
(BROAD BEAN) (PLANTS, EFFECT OF URANIUM ON)  
(PLANTS, EFFECT OF RADIUM ON)

L 30097-66 EWT(m)  
ACC NR: AP6012875

SOURCE CODE: UR/0205/66/006/002/0278/0283

AUTHOR: Vavilov, P. P.; Verkhovskaya, I. N.; Popova, O. N.; Kodaneva, R. P.

ORG: Komi Branch, AN SSSR, Syktyvkar (Komi filial AN SSSR); Institute of Biochemistry  
im. A. N. Bakh. AN SSSR, Moscow (Institut biokhimii AN SSSR)

TITLE: The depressant effect of small doses of ionizing radiation on growing plants

SOURCE: Radiobiologiya, v. 6, no. 2, 1966, 278-283

TOPIC TAGS: ionizing radiation, radiation plant effect, plant physiology, plant growth,  
gamma irradiation

ABSTRACT: In view of previous findings that the growth of *Vicia faba* is significantly delayed in areas with large deposits of uranium or radium, similar experiments were carried out over a 2-year period with spring wheat and spring barley grown in experimental plots under the influence of gamma radiation from U and Ra ore (radiation dose of 0.005 — 0.1 r/day). The height, internodes, dry weight, number of heads, and number of grains per head were measured in both experimental and control plots. Although radiation had no

Card 1/2

UDC: 58.039.1

Card 2/2



S/205/63/003/001/026/029  
E065/E485

AUTHORS: Vavilov, P.P., Verkhovskaya, I.N., Kondaneva, R.P.,  
Popova, O.N.

TITLE: The growth and development of *Vicia faba* L. under  
conditions of increased U and Ra content

PERIODICAL: Radiobiologiya, v.3, no.1, 1963, 132-138

TEXT: To elucidate the relative importance of chemical and radiation effects of radioactive substances on plants, bean seedlings were grown in pots containing soil to which U (concentration  $4 \times 10^{-7}\%$ ) and Ra ( $1.06 \times 10^{-7}\%$ ) had been added, while control plants were grown in pots containing normal turf soil; control and treated pots were placed in one of two trenches, one having normal background radioactivity (0.00004 r/hour) and the other with a radiation level of 0.002 r/hour, derived from U ( $10^{-2}\%$ ) and Ra ( $10^{-7}\%$ ) sources in the walls. Observations were made for several weeks on the growth and development of plants kept under the four different environments. The results showed that the U and Ra had an injurious effect on growth, retarding the upward growth of the stem and the formation of new leaves,  
Card 1/2

S/205/63/003/001/026/029  
E065/E485

The growth and development ...

reducing the general productivity (including fruit yield) and accelerating the processes of ageing. It was clearly shown that these effects were exerted when the U and Ra were not present in the soil and thus had no direct contact with the plants, indicating that their effects on growth are primarily associated with their radiation emissions rather than chemical properties. Analysis of plants grown on the supplemented soil indicated that Ra had been absorbed into the roots, leaves, stems and fruits, proving a source of chronic internal radiation. There was, however, evidence that chemical factors, particularly in the case of U, also influenced growth adversely in a distinctive manner. Reasons for the effects on growth of the very minute radiation doses applied in the investigation are discussed: the results are thought to indicate the need for a revision of the threshold doses currently accepted for higher plants. There are 2 figures and 3 tables.

ASSOCIATION: Laboratoriya radiobiologii Komi filiala AN SSSR,  
Syktyvkar (Radiobiology Laboratory, Komi Branch  
AS USSR, Syktyvkar)  
Card 2/2 SUBMITTED: May 7, 1962

POPOVA, O. N.

Gol'dberg, D. O., Sanamova, R. A., and Popova, O. N. "Deparaffination of  
(Surakhanskogo) residual oil in the presence of admixtures," Azerbaydzh. neft.  
khoz-vo, 1948, No. 11, p. 18-19

SO: U-3264, 10 April 53 (Letopis 'Zhurnal 'nykh Statey, No. 4, 1949).

Popova, O.N.

*Med* Effect of phosphate nutrition of tomatoes on the quality of the seed.  
A. A. Anisimov and O. N. Popova (*Agrobiologiya*, 1954, No. 5, 110—  
111).—Application of small amounts of P fertiliser (5 g. per 2 sq. m.  
of frame) improved seed quality. Seed from treated plants produced  
larger plants, earlier yields and fruit having higher sugar and  
vitamin C and lower acid contents. *Hort. Abstr.* (A. G. P.).

2

VAVILOV, P.P.; POPOVA, O.N.; KODANEVA, R.F.

Radium behavior in plants. Dokl. AN SSSR 157 no.4:992-994  
Ag '64 (MIRA 17:8)

1. Institut biologii Komi filiala AN SSSR. Predstavleno akademikom N.M. Sisakyanom.

DEGTYAROVA, A.S.[Dehtiar'ova, A.S.], kand. biol. nauk; PERESIPKIN, V.F.[Peresypkin, V.F.], doktor biol. nauk, prof.; BURKATSKAYA, O.M.[Burkata'ska, O.M.], kand. med. nauk; SPINU, I.E.I., kand. med. nauk; KOGAN, Yu.S.[Kohan, IU.S.], kand. med.nauk; MEYSAKHOVICH, Ya.O.[Meisakhovich, IA.O.], kand. tekhn. nauk; SANIN, V.A., kand. sel'khoz. nauk; TARANUKHA, M.D., kand. biol. nauk; ZAGOVORA, O.V.[Zahovora, O.V.], kand. sel'khoz. nauk; ARESHNIKOV, B.A.[Areshnykov, B.A., kand. biol. nauk; SALUNSKAYA, N.I.[Saluns'ka, N.I.], kand. biol. nauk; KOLOBOVA, A.M., kand. sel'khoz. nauk; KITITSYN, I.E.M.[Kitytsyn, I.E.M.], kand. sel'khoz. nauk; TELENGA, M.A.[Telenha, M.A.], doktor biol. nauk, prof.; POPOVA, O.O., kand. sel'khoz. nauk; MISKO, L.A.; ZRAZHEVSKIY, A.I.[Zrazhevs'kyi, A.I.], kand. sel'khoz. nauk; DYADECHKO, M.F., kand. sel'khoz. nauk; LOPATIN, V.M., kand. biol. nauk; MARKHASEVA, V.A.[Markhas'ova, V.A.], kand. sel'khoz. nauk; FEDOTOVA, K.M., kand. sel'khoz. nauk; TSELLE, M.O., kand. biol. nauk; VASIL'YEV, V.P.[Vasil'iev, V.P.], doktor biol. nauk, prof.; SHISHKOVA, M.I.[Shyshkova, M.I.], kand. sel'khoz. nauk; MARICHEK, Yu.Y.[Marychek, IU.I.], kand. sel'khoz. nauk; SHTERENBERG, P.M.[Shterenberh, P.M.], kand. sel'khoz. nauk; DMITRIYEV, Yu.V., kand. biol. nauk; LOZINSKIY, V.A.[Lozyns'kyi, V.A.], kand. biol. nauk; BRATUS', V.M., kand. sel'khoz. nauk; SOKUR, I.T., doktor biol. nauk; BOLOTIN, K.M.; SAVCHENKO, I.E.M., red.; NEMCHENKO, I.IU., tekhn. red.

(Continued on next card)

POPOVA, O. P., SELEGY, M. V., TUROVSKAYA, F. M., AMIKEYENKO, G. M.,  
GORYASHCHENKO, I. K., NUSBAUM, D. G., STEPANOVA, Z. P., LASOVA, O. I.,  
GALAKTIONOVA, N. S., GELKER, V. S., DENISEVA, O. A., YEMOKIMOVA, Y. I.,  
ZAKHAR'INA, R. A., KORCHNAYA, R. A., LEVINA, I. V., MISHCHENKO, E. B.,

"Hygienic characteristics of the day regimen of  
Moscow school children."

report submitted at the 13th All-Union Congress of Hygienists,  
Epidemiologists and Infectionists, 1959.





Papova, O. S.

✓ Mechanism of growth and the structure of metal precipitates formed during the electrocrystallization process.  
K. M. Gorbunova, O. S. Papova, A. A. Sutyagina, and Yu. M. Polukarov. *Russ. Kristallogr. Akad. Nauk S.S.S.R. Inst. Krist., Doklady Sovetskoye Akad. Nauk S.S.S.R.* 1956, 68-66 (Pub. 1957).—A discussion with 16 references. A. P. K.

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~~POPOVA, O. S.~~ POPOVA, O. S.

X-ray investigation of remanent stresses in electrolytic nickel deposits. V. P. Moiseev and O. S. Popova. *Izv. Akad. Nauk S.S.S.R., Ser. Fiz.* 20, 847-8 (1978).—Mat deposits, 20  $\mu$  thick, were obtained from a bath contg. 140 g./l.  $\text{NiSO}_4 \cdot 7\text{H}_2\text{O}$ , 30 g./l.  $\text{H}_2\text{BO}_3$ , 3 g./l.  $\text{NaCl}$ , 5 g./l.  $\text{NaF}$ , and bright deposits were obtained upon further addn. of 4 g./l. 2,6- or 2,7-naphthalene disulfonic acid at a c.d. of 1 amp./sq. dm. In bright deposits the initial strains are compression strains going over into tension strains. In mat deposits the original strains are tension strains which subside with increasing thickness. In bright Ni there are 8.3 vols. of absorbed H per vol. Ni, in mat deposits only 2.2 vols./vol. The tensions are relieved by heating to 800-900° with simultaneous release of H. An irregularly distributed interstitial soln. of H is postulated. S. Piskunov.

2  
Chem.

SM

POPOVA, O.S.

18  
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/ \*Internal Stresses in Electrodeposits. L. Stresses in Zinc Deposits. R. M. [unclear] and O. S. Popova (*Zhur Fiz Khim.*, 1955, 30 (2), 241) [in Russian]. After a bibliographical survey of the processes of stress formation and the influence of the surface [unclear] on the ductility of the metal, G. and P. describe the results of an investigation of the formation and nature of stresses in Zn deposits obtained from  $ZnSO_4$  soln. with and without addn. agents. The stress increases with increase of c.d., and at a given c.d. decreases with increase in thickness of the deposit in cases of simple  $ZnSO_4$  soln., but on addn. of dextrine this decrease is small, probably because of the  $d$  of the deposit and the presence of twinning. Crystallographic investigations of the deposits revealed the influence of the nature of the electrolyte on the directional properties of the crystal structure. The method of investigation was similar to that adopted by Samartsev (cf. *ibid.*, 1955, 29, 374; *M.A.*, 24, 132). 10 ref.—A. W.

fra 18  
day

L 05868-67 ENT(m)

ACC NR: AP6029973

SOURCE CODE: UR/0413/66/000/015/0166/0166

INVENTOR: Satalkin, A. V.; Popova, O. S.; Sokolovskiy, V. T.; Solntseva, V. A. 29  
B

ORG: none

TITLE: Method of preparing water-resistant concrete or solutions. Class 80, No. 184691  
/announced by Leningrad Order of Lenin Institute of Railroad-Transportation Engineers  
Im. Academician V. N. Obrastsov (Leningradskiy instituta inzhenerov zheleznodorozhnogo  
transporta)/

SOURCE: Izobret prom obraz tov zn, no. 15, 1966, 166

TOPIC TAGS: concrete, cement, material deformation, wear resistance

ABSTRACT: An Author Certificate has been issued for a method of preparing water-resistant concrete or solutions by mixing together cement, fillers, and water containing polymeric additives. Better water resistance and product deformation result when 1.5--2.5% by weight of polymeric additives, consisting of water-soluble epoxy resins (diethylene glycol or triethylene glycol) and a polyethylenepolyamine hardener, are introduced into the cement. [SA]

SUB CODE: 11, 07/ SUBM CODE: 25May64

KH

Card 1/1

UDC: 666.972.522:666.972.16

|   |   |
|---|---|
| L 13884-66 EWT(m)/EWP(j) WW/RM  |   |
| ACC NR: AP6005406 (A)   | SOURCE CODE: UR/0101/66/000/001/0014/0015 |
| AUTHOR: <u>Sataikin, A. V.</u> (Doctor of technical sciences); <u>Solntseva, V. A.</u> (Candidate of chemical sciences); <u>Popova, O. S.</u> (Engineer) 25   |   |
| ORG: <u>Leningrad Institute of Rail Transport Engineers</u> (Leningradskiy institut inzhenerov zheleznodorozhnogo transporta) 22 B  |   |
| TITLE: Cement with increased extensibility  |   |
| SOURCE: Tsement, no. 1, 1966, 14-15   |   |
| TOPIC TAGS: concrete, cement, reinforced concrete, synthetic resin additive, mechanical property  |   |
| ABSTRACT: A study has been made of the effect of synthetic resin additives on mechanical characteristics of portland cement and concrete to produce concrete of increased extensibility, suitable for road pavement, airfield runways surfacing etc. Comparative mechanical tests of fine grained concretes with new water-soluble additives, such as <u>TEG-17</u> and <u>DEG-1</u> epoxy resins and 89 resin, and with previously used additives, indicated a substantial increase in tensile and compressive <u>strength</u> and water resistance, and a 4—5-fold increase in extensibility of the portland cement with new water-soluble polymer additives. The highest tensile and compressive strength was determined in concretes with 2% additions of 89 and TEG-17 resins in aqueous medium. The high early strength increased further with increasing time of set under all setting |   |
| Card 1/2  | UDC: 666.958                              |

L 13884-66

ACC NR: AP6005406

3

conditions. The highest specific extensibility was also achieved with 89 resin. Deformability on compression and creep limit did not increase by addition of the water-soluble resins. The combination of increased extensibility with an increase in tensile strength of concrete may lead to an increase in crack resistance of concrete, i.e., in reliability and durability of concrete and reinforced concrete structures. Orig. art. has: 4 figures and 2 tables.

SUB CODE: 11/ SUBM DATE: none/ ORIG REF: 004/ ATD PRESS: 4/93

SB

Card 2/2

L 9432-66 EWT(m)/EWP(j) RM  
 ACC NR: AP5025439 SOURCE CODE: UR/0097/65/000/009/0029/0031  
 AUTHORS: Sataikin, A. V. (Doctor of technical sciences, Professor); Popova, O.  
S. (Engineer) 44, 55 45  
 ORG: none 15, 44, 55 B  
 TITLE: Strength and deformability of finely grained concretes with polymer additives  
 SOURCE: Beton i zhelezobeton, no. 9, 1965, 29-31  
 TOPIC TAGS: resin, construction material, polyvinyl acetate, plastic compound, polymer, concrete/ SKS 65GP latex, S 89 resin, DEG 1 resin, TEG 17 resin  
 ABSTRACT: The results of a study of the effect produced by polymer additives on the physico-mechanical and deformability properties of fine-grained concretes are presented. The following polymers were used in the tests: a polyvinyl acetate-water dispersion stabilized with formalin, divinyl styrol latex SKS-65GP, stabilized with nonionogenic soaps of type OP-7 and water-soluble resins S-89, DEG-1 and TEG-17. Stiff and plastic mixes were tested. The stiff mixes (105--110 mm slump on lab table vibrating at 2750 oscillations per minute for  
 Card 1/3 UDC: 620.17:666.97:691.175

L 9432-66

ACC NR: AP5025439

20--30 seconds) were found significantly superior. Figure 1 shows a plot of the

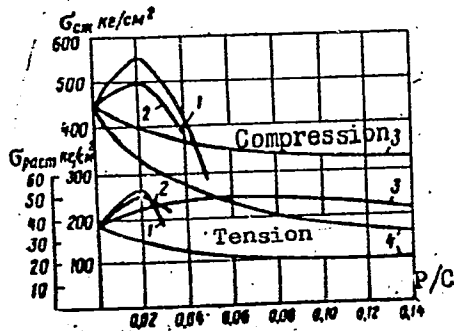


Fig. 1. Variation of compressive and tensile strength limits with the P/C (polymer-cement) ratio for a 28-day duration. 1 - resin 89 additive; 2 - resin TEG-17 additive; 3 - latex SKS-65GP additive; 4 - polyvinyl acetate additive stabilized in formalin.

relationship of the compressive and tensile strength limits at varying polymer-cement ratios. It is noted that the strength limits are larger for specimens having water-soluble resin additives. An optimal quantity of a polymer additive is defined as that quantity which increases both the compressive and tensile strength limits. A relationship was found for the variation of compressive and tensile strength with duration of specimen immersion in water. The experiments

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ACC NR: AP5025439

0

indicated a beneficial effect of the polymer additives on concrete hydration. The polymer-bearing mixes were tested for creep and deformation characteristics and were compared with the same characteristics as exhibited by indicated mixes. The results of the experimental measurements are tabulated. The resin additives DEG-1, TEG-17 and S-89 yielded favorable properties of mixability and strength without appreciably lowering the compressive deformation characteristics. Orig. art. has: 4 tables and 5 figures.

SUB CODE: 11/

SUBM DATE: none/

ORIG REF: 002

Card 3/3

S/564/57/000/000/003/029  
D258/D307

AUTHORS:

Gorbunova, K. M., Popova, O. S., Sutyagina, A.  
A., and Polukarov, Yu. M.

TITLE:

Mechanism of growth and structure of metallic  
deposits formed during electrocrystallization

SOURCE:

Rost kristallov; doklady na Pervom soveshchanii  
po rostu kristallov, 1956 g. Moscow, Izd-vo  
AN SSSR, 1957, 58-66

TEXT: The present article is a review of some of the authors' earlier studies and other work; attention is focused on some regularities concerning the cathode deposition of metals, as dense or porous, dendritic, coatings. An increase in the electrolysis current leads to an increase in the surface of the crystallizing metal, leading to an increase in the number of crystals or to dendritization, according to conditions. Powder deposition (fine dendrites) occurs at the saturation current.

Card 1/2

Mechanism of growth...

S/564/57/000/000/003/029  
D256/D307

Properties of such polydendritic and of dense galvanic coatings are indicated, and conditions leading to the formation of electrolytic deposits with a predetermined crystal orientation are discussed with particular reference to the earlier work of Gorbunova. It is believed that orientation may be ascribed to the unequal rates of growth of variously oriented crystals. Possible mechanisms for the formation of texture on coatings are indicated. An account is also given of the authors' earlier study of internal stresses in galvanic coatings, particularly in the presence of surface-active compounds, and of the conditions leading to coatings possessing difference surfaces. There are 6 figures and 16 references: 12 Soviet-bloc and 4 non-Soviet-bloc.

Card 2/2

SANZHAROVSKIY, A.T.; POPOVA, O.S. (Moscow)

Study of internal stresses arising in cathodic hydrogen  
absorption by metals. Zhur.fiz.khim. 35 no.11:2646-2648 1961.  
(MIRA 14:12)

1. Akademiya nauk SSSR, Institut fizicheskoy khimii.  
(Metals--Hydrogen content)

POPOVA, O.S.; SANZHAROVSKIY, A.T.

Some data on the effect of cathodically reduced hydrogen on  
the properties of iron and nickel. Zhur.prikl.khim. 34 no.9:  
2117-2120 S '61. (MIRA 14:9)

1. Institut fizicheskoy khimii AN SSSR.  
(Iron) (Nickel) (Hydrogen)

SANZHAROVSKIY, A.T.; POPOVA, O.S.

Some data on the effect of cathodically reduced hydrogen on  
the properties of electrodeposited nickel. Zhur.prikl.khim.  
34 no.9:2120-2123 S '61. (MIRA 14:9)

1. Institut fizicheskoy khimii AN SSSR.  
(Nickel plating) (Hydrogen)

MOISEYEV, V.P.; POPOVA, O.S.

X-ray investigation of phase transformations in the electrolytic precipitates of manganese. Fiz. met. i metalloved. 9 no. 4:584-588 Ap '60. (MIRA 14:5)

1. Institut fizicheskoy khimii AN SSSR.  
(Manganese—Metallography) (Electrolysis)

18.7530

27348  
S/080/61/034/009/014/016  
D204/D305

AUTHORS: Popova, O.S., and Sanzharovskiy, A.T.

TITLE: Certain data on studying the effect of cathodically reduced hydrogen on the properties of iron and nickel

PERIODICAL: Zhurnal prikladnoy khimii. v. 34, no. 9, 1961,  
2117 - 2120

TEXT: Communication II of a series of investigations into the effect of cathodically reduced hydrogen on the properties of metals. A study has been made on the effect of hydrogen on electrolytically-produced bi-metal foils (of iron and nickel). If the iron-nickel diaphragm is hydrogenated, then the coating (Ni) has practically no effect on the diffusion of the hydrogen. If, however, hydrogenation takes place on the nickel side then the diffusion time is considerably extended. Freyman and Titov's method was used to determine the quantity of hydrogen passing through a metal layer polarized on one side. Kinetics of hydrogenation and determination

Card 1/5



Certain data on studying ...

27328  
S/080/61/034/CO9 014/1 8  
D204/D305

of the internal stresses set up by the hydrogen were studied by the flexible cathode method. Having assumed that the depth of penetration of hydrogen and stresses due to it do not depend on the thickness of the foil, then the magnitude of the bend of hydrogenated foil will be inversely proportional to its toughness. From two experiments using foils of different thickness the depth of penetration of hydrogen was calculated and the stresses set up could be calculated from the equation:

$$\sigma = \frac{Et^3h}{3l^2(xt - x^2)}$$

where  $\sigma$  - stress,  $E$  - modulus of elasticity,  $l$  - length of the foil  
 $h$  - declination of the lower end of the bend of the cathode. The hydrogen occluded in the metal was determined by pumping out in a vacuum. Foils saturated with hydrogen were prepared by cathode polarization in a 10 %  $H_2SO_4$  solution with 0.1 g/l.  $Na_2SO_4$  added at a current density of 100 mA/cm<sup>2</sup> and temperature 20-30°. Fig. 1

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27348

S/080/61/034/009/014/016

D204/D305

Certain data on studying ...

(curve 1) shows that hydrogen penetrates practically instantaneously through the foil and the rate of diffusion remains constant. On switching off the polarizing current, hydrogen continues to be evolved from the foil, giving evidence of the large amount of hydrogen dispersed in the metal. After passing hydrogen through the iron foil for 30-40 minutes blisters and cracks were discovered, these surface effects being largely of a local character. Hydrogenation causes reduced strength - from 50 to 40 kg/cm<sup>2</sup>. X-ray examination showed no changes in the metal lattice. As for nickel (Fig. 1, curve 2), after switching off the polarizing current, the system returns to its original condition. Fig. 4 indicates the kinetics of a nickel foil when cathodically polarized. Analysis of the curve shows that change of volume of the system is not connected with penetration of hydrogen, but with the bend which occurs on switching off the polarizing current and with the escape of hydrogen from the metal. Directly after the experiment the nickel was found to be brittle, but on pumping out the hydrogen, the brittleness disappears and the breaking strain is raised from 34 to 38

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27346

S/080/61/034/009/014-002

D204/D305

Certain data on studying ...

kg/cm<sup>2</sup>. When hydrogen passes through the whole thickness of iron foil (100-500  $\mu$ ) there are no significant changes in the lattice. Occlusion of hydrogen in iron causes internal stresses and local destruction and irreversible lowering of strength. When nickel is cathodically polarized, hydrogen penetrates to a depth of the order 20-25  $\mu$  creating internal stresses of 10 kg/mm<sup>2</sup>. Hydrogen diffusion brings about a break-up of grain blocs and brittleness. The stresses are removed after 1-1.5 hours and the brittleness after 60-70 hours from the time of switching off the polarized current. Hydrogen is automatically evolved as the internal stress is removed. After all the hydrogen has escaped, the strength is somewhat higher than the original, which seems to be conditioned by the dispersion of the grains. There are 5 figures, and 6 references: 5 Soviet-bloc and 1 non-Soviet-bloc. The reference to the English-language publication reads as follows: B. Baranovskiy, and M. Smelovskiy, The Physics and Chemistry of Solids, 12, 2, 1960. ASSOCIATION: Institut fizicheskoy khimii AN SSSR (Institute of Physical Chemistry AS USSR)

SUBMITTED: June 23, 1960

Card 4/5

18.7530 1416 1413 1418

27349

S/080/61/034/009/015/016

D204/D305

AUTHORS: Sanzharovskiy, A.T., and Popova, O.S.

TITLE: Some data on studying the effect of cathodically reduced hydrogen on the properties of electrolytic nickel

PERIODICAL: Zhurnal prikladnoy khimii, v. 34, no. 9, 1961, 2120 - 2123

TEXT: This is communication II of a series of investigations into the effect of cathodically reduced hydrogen on the properties of metals. Hydrogen diffusion through an electrolytically produced nickel diaphragm is a difficult subject to study because the metal so readily becomes embrittled, so experiments were made by the flexing cathode method, using two-layer samples (the nickel deposits were cemented to rolled nickel sheet or foil with cement BF-2). A relationship was obtained between the bend of the foil  $h$  and time  $\tau$ . The curve has 3 sections -- section 0-1, intensive hydrogenation of the deposit accompanied by increase of internal  
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273h9  
S/080/61/034/009/015/016  
D204/D305

Some data on studying the ...

stress; cracks appear in the deposit at point 1 and cause the stress to fall off. At point 2 the polarization current was switched off and section 2-3 indicates release of hydrogen from the metal and relief of internal stress, which, experiments showed, was entirely removed. This indicates that there is no deformation of the metal, but after the stresses have been removed the deposits are still very brittle, their toughness is around zero. However, holding these deposits at room conditions for 60-70 hours causes the brittleness entirely to disappear and the breaking strain is even higher than the original (109 kg/mm<sup>2</sup> as compared with 88-90 kg/mm<sup>2</sup> for bright deposits, and 40 kg/mm<sup>2</sup> as compared with 34 kg/mm<sup>2</sup> for matt deposits). Hydrogen occluded by the deposits was found to be 0.02 cm<sup>3</sup>/gm., and this is only removed by heating to 100°. X-ray tests showed that the lines are more intense after hydrogenation than before, which may be due either to disruption of the grains and increased dispersion or to microdistortions of the structure. An attempt was made at determining the value of internal stress causing cracking, and for this two-layer foils were used,

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273h9

S/080/61/034/009/015/016

D204/D305

Some data on studying the ...

consisting of deposit thickness 10, 20, 30  $\mu$  cemented to nickel foil thickness 100  $\mu$ . To calculate the value of the internal stresses the formula used was:

$$\sigma = \frac{E(t + \Delta th)}{3l\Delta t}$$

where h - declination of lower end of foil, E - modulus of elasticity of the sub-layer,  $\Delta t$ , t - thickness of sublayer and layer respectively, l - working length of the cover (deposit). Calculations showed that bright deposits of thickness 10  $\mu$  crack at stress 5.3 kg/mm<sup>2</sup>; 20  $\mu$  at 10 kg/mm<sup>2</sup>; and 30  $\mu$  at 12 kg/mm<sup>2</sup>. The figures for matt deposits are somewhat smaller. The mechanism of the processes taking place may be thus briefly explained: the hydrogen diffusing through the thickness of the metal is dispersed in the cavity defects of the structure and creates a pressure of the order of 10 kg/mm<sup>2</sup>; another part of the hydrogen is adsorbed on the surface of these defects and lowers their surface energy; when the strength of the metal will have been reduced to the level of the

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27349

S/080/61/034/009/01/01  
D204/D305

Some data on studying the ...

internal stresses produced then the metal begins to break into pieces. Running the polarizing current for 40-60 minutes and then switching off causes complete removal of internal stress, although the deposit still remains brittle due to the effect of the adsorbed hydrogen. The brittleness can be removed by allowing the hydrogen to pass out of its own accord over a period of 60-70 hours or by placing the metal in a vacuum without heat for 15-20 minutes. The conclusion drawn from Communications I and II is that the interaction of iron and hydrogen is analogous to that of nickel and hydrogen, but whereas there is local destruction of iron and its strength remains considerable, the adsorption effect of hydrogen on nickel is much less than on iron. There are 5 figures and 3 Soviet-bloc references. X

ASSOCIATION: Institut fizicheskoy khimii AN SSSR (Institute of Physical Chemistry, AS USSR)

SUBMITTED: June 23, 1960

Card 4/4

POPOVA, O.S.; SAKZHAROVSKIY, A.T.

Effect of cathode-reduced hydrogen on the properties of metals.  
Dokl. AN SSSR 136 no. 2:654-656 Ja '61. (MIRA 14:2)

1. Institut fizicheskoy khimii AN SSSR. Predstavleno akademikom  
P.A. Rebinderom.  
(Hydrogen) (Metals--Hydrogen content)



SANZHAROVSKIY, A.T.; POPOVA, O.S.

Method for determining the diffusion of cathode-reduced hydrogen  
through metals. Zhur. fiz. khim. 34 no. 11:2601-2602 N '60.  
(MIRA 14:1)

1. Akademiya nauk SSSR, Institut fizicheskoy khimii.  
(Hydrogen) (Diffusion)

89735

18.8200

S/020/61/136/003/023/027  
B004/B056

AUTHORS: Popova, O. S. and Sanzharovskiy, A. T.

TITLE: Effect of Cathode-reduced Hydrogen on the Properties of Metals

PERIODICAL: Doklady Akademii nauk SSSR, 1961, Vol. 136, No. 3, pp. 654-656

TEXT: The effect of cathode-reduced hydrogen upon the mechanical properties of metals is studied. For this purpose, the effect produced by cathodic polarization upon rolled iron, rolled nickel and electrodeposited nickel is studied. The cathodic polarization was carried out in a 10% sulfuric acid with an addition of 0.1 g/l Na<sub>2</sub>S, current density 100 ma/cm<sup>2</sup>, temperature 20 - 25°C. The following measurements were made: 1) Measurement of the diffusion of H<sub>2</sub> into the metal; 2) Measurement of the deformation of unilaterally polarized lamellas due to H<sub>2</sub> adsorption; 3) Determination of the adsorbed H<sub>2</sub> by means of extraction in vacuum; 4) X-ray analysis; 5) Determination of the strength before and after cathodic polarization. X

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89735

Effect of Cathode-reduced Hydrogen on the  
Properties of Metals

S/020/61/136/003/023/027  
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The following results are enumerated. A) Rolled nickel: Hydrogen diffuses only to a depth of  $30\mu$ ; causing internal stress of the magnitude of  $10 \text{ kg/mm}^2$ . Brittleness occurs, and strength is decreased. This brittleness disappears after 60 - 70 hours of keeping the specimen in air. During this time the entire occluded hydrogen is eliminated. The strength increases as compared to the initial value by 5-6%. B) Electrodeposited nickel: Hydrogen diffuses deeper into the metal than in the case of rolled nickel, and in polished surfaces more quickly than in dim ones. Apart from the fact that in electrodeposited nickel cracks easily occur due to occluded hydrogen, the behavior is analogous to A). C) Rolled iron: Hydrogen penetrates very deep into the metal. Internal stress of about  $15 \text{ kg/mm}^2$ , increased brittleness, and local destruction occur. The major quantity of occluded hydrogen is eliminated completely only after 6 - 7 days, the strength, however, remains irreversible (15-20%), and flexibility is reduced by 50%. A change in the lattice parameters was, however, not observed. The following conclusions are drawn: The  $\text{H}_2$  penetrated into the metal collected in the microcavities of structural defects and thereby causes internal stress. Part of the  $\text{H}_2$  is adsorbed on the surface of the

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Effect of Cathode-reduced Hydrogen on the  
Properties of Metals

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defects, reduces the surface energy and thus also the strength of the metal. While in the diffusion of  $H_2$  into Ni the effect of the adsorptive reduction of strength predominates, in the diffusion of  $H_2$  into Fe that of the effect of the internal stress predominates. Experiments showed that accelerated extraction of hydrogen in the vacuum also eliminated brittleness. This proves the connection between brittleness and hydrogen content. There are 4 figures and 5 references: 4 Soviet and 1 Polish.

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